

# Teknisk specifikation

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### **Nickellegeringar – Bestämning av nickelinnehåll – Optisk emissionsspektrometri med induktivt kopplat plasma**

### **Nickel alloys – Determination of Nickel content – Inductively coupled plasma atomic emission spectrometric method**

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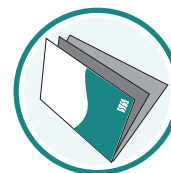
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## Foreword

ISO (the International Organization for Standardization) is a worldwide federation of national standards bodies (ISO member bodies). The work of preparing International Standards is normally carried out through ISO technical committees. Each member body interested in a subject for which a technical committee has been established has the right to be represented on that committee. International organizations, governmental and non-governmental, in liaison with ISO, also take part in the work. ISO collaborates closely with the International Electrotechnical Commission (IEC) on all matters of electrotechnical standardization.

The procedures used to develop this document and those intended for its further maintenance are described in the ISO/IEC Directives, Part 1. In particular the different approval criteria needed for the different types of ISO documents should be noted. This document was drafted in accordance with the editorial rules of the ISO/IEC Directives, Part 2 (see [www.iso.org/directives](http://www.iso.org/directives)).

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For an explanation on the meaning of ISO specific terms and expressions related to conformity assessment, as well as information about ISO's adherence to the WTO principles in the Technical Barriers to Trade (TBT) see the following URL: [Foreword - Supplementary information](#)

The committee responsible for this document is ISO/TC 155, *Nickel and nickel alloys*.

# Nickel alloys — Determination of Nickel content — Inductively coupled plasma atomic emission spectrometric method

## 1 Scope

This Technical Specification describes an inductively coupled plasma atomic emission spectrometric method for the determination of nickel content (mass fraction) between 20,0 % and 80,0 % in nickel alloys.

Besides alloys where nickel is the main component regarding its content level, this method may also apply to alloys in which nickel has a content as high as several other elements (Fe, Cr, Co,...) and in which the “main element” cannot be specified.

## 2 Normative references

The following documents, in whole or in part, are normatively referenced in this document and are indispensable for its application. For dated references, only the edition cited applies. For undated references, the latest edition of the referenced document (including any amendments) applies.

ISO 648, *Laboratory glassware — Single-volume pipettes*

ISO 1042, *Laboratory glassware — One-mark volumetric flasks*

ISO 3696, *Water for analytical laboratory use — Specification and test methods*

## 3 Principle

Dissolution of a test portion in a mixture of hydrofluoric, hydrochloric, nitric and phosphoric acid, and fuming after addition of perchloric acid.

If necessary, addition of extra hydrofluoric acid and, if desired, addition of an internal reference element (scandium recommended).

Dilution of the test solution to a known volume. Nebulisation of this solution into an inductively coupled plasma atomic emission spectrometer and measurement of the intensity of the emitted light (including, where appropriate, that of the internal reference element).

The method uses a calibration based on a very close matrix matching of the calibration solutions to the sample composition and a bracketing of the mass fractions between  $\pm 2$  % of the approximate content of nickel in the sample to be analysed.

The content of all elements in the sample has, therefore, to be approximately known. If the composition is not known, the sample shall be analysed by some semi quantitative method.

The advantage with this procedure is that all possible matrix interferences will be minimized which will result in a higher trueness. This is important for spectral interferences which can be severe in very highly alloyed matrixes. All possible interferences shall be kept at a minimum level. Therefore, it is essential that the spectrometer used meets the performance criteria recommended for the selected analytical lines.

The wavelengths reported in [Table 1](#) have been investigated and the strongest possible interferences are given. If other wavelengths are used, they shall be carefully checked. The wavelength for the internal reference element should be selected carefully. The use of scandium at 363,07 nm is recommended. This line is interference-free for the elements and contents generally found in nickel alloys.

**Table 1 — Examples of wavelengths for Nickel**

Element	Wavelengths, nm	Interferences
Nickel	231,604	Co, Mo
	230,299	/
	222,486	Co
	222,295	V, Co
	217,514	/
	227,877	/
	239,452	/
	221,647	Co
	227,021	/

## 4 Reagents

During the analysis, use only reagents of recognized analytical grade and only “grade 2” water as specified in ISO 3696, or water of equivalent purity.

The same reagents should be used for the preparation of calibration solutions and of sample solutions.

**4.1 Hydrofluoric acid, HF, 40 % (m/m),  $\rho = 1,14$  g/ml, or 50 % (m/m),  $\rho = 1,17$  g/ml.**

**WARNING — Hydrofluoric acid is extremely irritating and corrosive for the skin and the mucous membranes producing severe skin burns which are slow to heal. In the case of contact with skin, wash well with water, apply a topical gel containing 2,5 % (mass fraction) calcium gluconate, and seek immediate medical treatment.**

**4.2 Hydrochloric acid, HCl,  $\rho = 1,19$  g/ml.**

**4.3 Nitric acid, HNO<sub>3</sub>,  $\rho = 1,40$  g/ml.**

**4.4 Phosphoric acid, H<sub>3</sub>PO<sub>4</sub>,  $\rho = 1,70$  g/ml.**

**4.5 Perchloric acid, HClO<sub>4</sub>, 60 % (m/m)  $\rho = 1,54$  g/ml or 70 % (m/m),  $\rho = 1,67$  g/ml.**

**WARNING — Perchloric acid vapour may cause explosion in the presence of ammonia, nitrous fume, or organic matter, in general. All evaporation shall be carried out by using fume hoods specifically designed for perchloric acid evaporations.**

**4.6 Internal reference element solution, 100 mg/l.**

Choose a suitable element to be added as internal reference and prepare a 100 mg/l solution.

NOTE 1 Commercially available standard solutions can be used for this purpose.

NOTE 2 Scandium was used during the precision test of this method.

**4.7 Nickel standard solution, 10 g/l.**

Weigh  $5 \text{ g} \pm 0,001 \text{ g}$  of high purity nickel [min 99,9 % (m/m)], transfer it to a beaker and dissolve in 50 ml of water and 100 ml of nitric acid (4.3). Cover with a watch glass and heat gently until the nickel is completely dissolved. Cool and transfer quantitatively into a 500 ml one-mark volumetric flask. Dilute to the mark with water and mix.



1 ml of this solution contains 10 mg of nickel.

NOTE Commercially available standard solutions are not allowed.

#### **4.8 Nickel standard solution, 5 g/l.**

Weigh  $5 \text{ g} \pm 0,001 \text{ g}$  of high purity nickel [min 99,9 % (m/m)], transfer it to a beaker, and dissolve in 50 ml of water and 100 ml of nitric acid (4.3). Cover with a watch glass and heat gently until the nickel is completely dissolved. Cool and transfer quantitatively into a 1 000 ml one-mark volumetric flask. Dilute to the mark with water and mix.

1 ml of this solution contains 5 mg of nickel.

NOTE Commercially available standard solutions are not allowed.

#### **4.9 Standard solutions of matrix elements.**

Prepare standard solutions for each element whose content (mass fraction) is higher than 1 % in the sample. Use pure metals or chemical substances with nickel content (mass fraction) less than 100 µg/g.

## **5 Apparatus**

All volumetric glassware shall be class A and calibrated in accordance with ISO 648 or ISO 1042, as appropriate.

### **5.1 Polytetrafluoroethylene (PTFE) beakers or Perfluoroalkoxy-polymer (PFA) beakers with a graphite base.**

**5.2 Polypropylene or polyethylene volumetric flasks**, of capacity 250 ml, calibrated according to ISO 1042.

### **5.3 Optical emission spectrometer, equipped with inductively coupled plasma.**

This shall be equipped with a nebulisation system resistant to hydrofluoric acid. The instrument used will be satisfactory if, after adjustment according to 7.6, it meets the performance criteria given in 5.3.

The spectrometer can be either a simultaneous or a sequential one. If a sequential spectrometer can be equipped with an extra device or simultaneous measurement of the internal reference element line, it can be used with the internal reference method. If the sequential spectrometer is not equipped with this device, an internal reference cannot be used and an alternative technique without an internal reference element shall be used.

#### **5.3.1 Practical resolution of a sequential spectrometer.**

Calculate the bandwidth (full width at half maximum), according to A.1, for the line used, including that for the internal reference element. The bandwidth shall be less than 0,030 nm.

#### **5.3.2 Short-term stability.**

Calculate the standard deviation of 10 measurements of the absolute intensity or intensity ratio corresponding to nickel and to the internal reference element, by using the most concentrated calibration solution for nickel, according to A.2. The relative standard deviation should not exceed 0,2 %.

NOTE Short-term stability values from 0,12 % to <0,8 % were obtained during the precision test of this method.

### 5.3.3 Background equivalent concentration.

Calculate the background equivalent concentration (BEC) according to [A.3](#), for the analytical line, using a solution containing only the analyte (nickel). Maximum BEC values should be less than 0,4 mg/l.

NOTE Background equivalent concentration values between 0,045 mg/l and 0,30 mg/l were obtained during the precision test program.

## 6 Sampling and sample preparation

Sampling and sample preparation shall be carried out by normal agreed procedures.

The laboratory sample is normally in the form of millings or drillings and no further mechanical preparation is necessary.

The laboratory sample shall be cleaned by washing with pure acetone and drying in air.

If brazed alloy tools are used in the preparation of the laboratory sample, then the sample shall be further cleaned by pickling in 15 % (m/m) nitric acid for a few minutes. It shall then be washed several times with distilled water, followed by washing in acetone and drying in air.

## 7 Procedure

### 7.1 Sample composition

The approximate chemical composition of each sample (nickel content and all other matrix elements which content is above 1 %) shall be known preliminarily.

### 7.2 Test portion

Weigh, to the nearest 0,001 g, 0,25 g of the test sample.

### 7.3 Preparation of test solution, $T_{Ni}$

A PTFE or PFA beaker ([5.1](#)) should be used when using HF ([4.1](#)).

**7.3.1** Transfer the test portion ([7.2](#)) into a 250 ml PTFE or PFA beaker with a graphite base ([5.1](#)).

**7.3.2** Add 5 ml HF ([4.1](#)), 30 ml HCl ([4.2](#)), and 3 ml HNO<sub>3</sub> ([4.3](#)) and allow the sample to dissolve overnight at room temperature. Add 2,5 ml H<sub>3</sub>PO<sub>4</sub> ([4.4](#)). If necessary, heat gently to complete dissolution. Add 7,5 ml HClO<sub>4</sub> ([4.5](#)) and heat until the perchloric acid starts to fume. Continue fuming for two to three minutes.

NOTE 1 Other HF and HCl volumes can be used, depending on the composition of the sample.

NOTE 2 5 ml H<sub>3</sub>PO<sub>4</sub> ([4.4](#)) and 5 ml HClO<sub>4</sub> ([4.5](#)) can also be used, instead of 2,5 ml H<sub>3</sub>PO<sub>4</sub> ([4.4](#)) and 7,5 ml HClO<sub>4</sub> ([4.5](#)).

**7.3.3** Cool the solution and add 10 ml of water to dissolve the salts. Some residues may remain undissolved. In this case, add 2 ml of HF ([4.1](#)) and heat gently in a water bath until the residues dissolve completely.

**7.3.4** Cool the solution to room temperature and transfer it quantitatively into a 250 ml polypropylene or polyethylene volumetric flask labelled  $T_{Ni}$ . If internal reference element ([4.6](#)) is used, add 10 ml of this internal standard.

NOTE 3 Depending on the instrument performances, the volume and/or the concentration of the internal reference element solution may be different.

**7.3.5** Dilute to the mark with water and mix.

#### **7.4 Preparation of rinsing solution, T<sub>0</sub>**

In a 250 ml polypropylene or polyethylene volumetric flask (5.2) labelled T<sub>0</sub>, add 2,5 ml H<sub>3</sub>PO<sub>4</sub> (4.4), 7,5 ml HClO<sub>4</sub> (4.5), and 10 ml of the internal reference element (4.6). Dilute to mark with water and mix.

#### **7.5 Preparation of calibration solutions for bracketing: T<sub>l,Ni</sub> and T<sub>h,Ni</sub>**

For each test solution, T<sub>Ni</sub>, prepare two matrix matched calibration solutions, T<sub>l,Ni</sub> and T<sub>h,Ni</sub>, with nickel concentrations in T<sub>l,Ni</sub> slightly below and in T<sub>h,Ni</sub> slightly above the nickel concentration in the unknown test solution as follows.

**7.5.1** Add the suitable volume, K<sub>l,Ni</sub>, of the nickel standard solutions (4.7 or 4.8) in a 250 ml PTFE or PFA beaker (5.1) marked T<sub>l,Ni</sub> so that the mass of nickel, expressed as K<sub>l,Ni</sub> %, is approximately 2 % below the estimated mass content of nickel, expressed as K<sub>Ni</sub> %, in the sample.

The selected volume, K<sub>l,Ni</sub>, shall be easy to pipette.

**7.5.2** Add the suitable volume, K<sub>h,Ni</sub>, of the nickel standard solutions (4.7 or 4.8) in a 250 ml PTFE or PFA beaker (5.1) marked T<sub>h,Ni</sub> so that the mass of nickel, expressed as K<sub>h,Ni</sub> %, is approximately 2 % above the estimated mass content of nickel, expressed as K<sub>Ni</sub> %, in the sample.

The selected volume, K<sub>h,Ni</sub>, shall be easy to pipette.

**7.5.3** Add to the calibration solutions, T<sub>l,Ni</sub> and T<sub>h,Ni</sub>, all matrix elements having mass fractions above 1 % in the test solution using the appropriate volumes of standard solutions (4.9) to match the equivalent matrix composition to the nearest percent.

NOTE The calibration solutions, T<sub>l,Ni</sub> and T<sub>h,Ni</sub>, are specific for each sample.

**7.5.4** Add in each beaker, T<sub>l,Ni</sub> and T<sub>h,Ni</sub>, 2,5 ml H<sub>3</sub>PO<sub>4</sub> (4.4), 7,5 ml HClO<sub>4</sub> (4.5) and heat until the perchloric acid starts to fume. Continue to fume for two to three minutes.

**7.5.4** Continue as specified from 7.3.3 to 7.3.5.

#### **7.6 Adjustment of the apparatus**

**7.6.1** Start the inductively coupled plasma optical emission spectrometer and let it stabilize in accordance with the manufacturer's instructions.

**7.6.2** At one of the wavelengths listed in Table 1, adjust all appropriate instrumental parameters, as well as the pre-spraying and the integrating times, according to the instrument manufacturer's instructions while aspirating the highest concentration calibration solution.

Depending on the instrument configuration, these parameters may include the outer, intermediate or central gas flow-rates, the torch position, the entrance slits, the exit slits and the photomultiplier tubes voltage.

NOTE Other wavelengths can be used provided that interferences, sensitivity, resolution, and linearity criteria have been carefully investigated.